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Hydrocarbons Answer Chemistry Hydrocarbons Study Guide

Answers A controlled process by which hydrocarbons are broken down or rearranged into smaller, more useful molecules natural gas This is typically made up of 80% methane, 10% ethane, 4% propane, and 2% butane Study Guide: Chapter 22 Page 11/24

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Cyclohexane contains only single bonds and is representative of
the simplest type of cyclic hydrocarbons. A ring structure may
possess double bonds, as in the following portrayal of the
well-known hydrocarbon benzene, which has the composition C 6
H 6. See Figure 8. Figure 8. Benzene.

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Guide Key Concepts 22.1 Hydrocarbons • Because carbon has

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four valence electrons, carbon atoms always form four covalent bonds. • The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches. • Molecules of hydrocarbons, such as alkanes, are nonpolar molecules. 22.2 Unsaturated Hydrocarbons

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Chemists called the hydrocarbons that reacted with bromine unsaturated hydrocarbons in the same sense that an unsaturated aqueous solution can dissolve more solute. Hydrocarbons that did not react with bromine were said to be saturated. Present-day chemists can now explain the experimental results obtained 170 years ago.

Chapter 21: Hydrocarbons

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