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Types of Reactions Worksheet -

Solutions 1) $3 \text{NaBr} + 1 \text{H}_3\text{PO}_4 + 1 \text{Na}_3\text{PO}_4 + 3 \text{HBr}$ Reaction type: double displacement
2) $3 \text{Ca(OH)}_2 + 1 \text{Al}_2(\text{SO}_4)_3 + 3 \text{CaSO}_4 + 2 \text{Al(OH)}_3$ Reaction type: double displacement
3) $3 \text{Mg} + 1 \text{Fe}_2\text{O}_3 \rightarrow 2 \text{Fe} + 3 \text{MgO}$

Types of Reactions Worksheet

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Types of Reactions Worksheet -
Solutions Balance the following
equations and indicate the type of
reaction taking place: 1) $3 \text{NaBr} + 1$
 $\text{H}_3\text{PO}_4 + 1 \text{Na}_3\text{PO}_4 + 3 \text{HBr}$ Type of
reaction: double displacement 2) 3
 $\text{Ca(OH)}_2 + 1 \text{Al}_2(\text{SO}_4)_3 \rightarrow 3 \text{CaSO}_4 + 2$
 Al(OH)_3 Type of reaction: double
displacement 3) $3 \text{Mg} + 1 \text{Fe}_2\text{O}_3 \rightarrow 2 \text{Fe} +$

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3 MgO Type of reaction: single displacement

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main things we will present to you based on the ...

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Types of Chemical Reactions Answers
Balance each of the following reactions
and identify each type of reaction: 1. $2 \text{NaBr} + \text{Ca(OH)}_2 \rightarrow \text{CaBr}_2 + 2 \text{NaOH}$

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double displacement 2. $2 \text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow (\text{NH}_4)_2\text{SO}_4$
synthesis 3. $4 \text{C}_5\text{H}_9\text{O} + 29 \text{O}_2 \rightarrow 20 \text{CO}_2 + 18 \text{H}_2\text{O}$
combustion 4. $3 \text{Pb} + 2 \text{H}_3\text{PO}_4 \rightarrow 3 \text{H}_2 + \text{Pb}_3(\text{PO}_4)_2$
single displacement 5. $\text{Li}_3\text{N} + 3 \text{NH}_4\text{NO}_3 \rightarrow 3 \text{LiNO}_3 + (\text{NH}_4)_3\text{N}$

Types of Chemical Reaction Worksheet

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Five Types of Chemical Reaction
Worksheet Answers. Balance the
following reactions and indicate which of
the six types of chemical reaction are
being represented: a) $2. \text{NaBr} + 1. \text{Ca(OH)}_2 \rightarrow 1. \text{CaBr}_2 + 2. \text{NaOH}$ Type of
reaction: double displacement. b) $2. \text{NH}_3 + 1. \text{H}_2\text{SO}_4 \rightarrow 1. (\text{NH}_4)_2\text{SO}_4$ Type of
reaction: synthesis. c) $4. \text{C}_5\text{H}_9\text{O} + 27.$

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O₂ (20. CO₂ + 18. H₂O Type of reaction:

Five Types of Chemical Reaction Worksheet Answers

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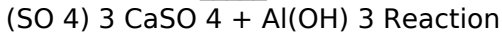
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Types of Reactions Worksheet Indicate the type of reaction taking place: 1)



Reaction type: _____ 2) $\text{Ca}(\text{OH})_2 + \text{Al}_2$



Reaction type: _____ 3) $\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow \text{Fe} + \text{MgO}$

Reaction type: _____ 4) $\text{C}_2\text{H}_4 + \text{O}_2$

Types of Reactions Worksheet -

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Types of Reactions Worksheet - Solutions. Balance the following equations and indicate the type of reaction taking place: 1) $3. \text{NaBr} + 1. \text{H}_3\text{PO}_4 \rightarrow 1. \text{Na}_3\text{PO}_4 + 3. \text{HBr}$. Type of reaction: double displacement. 2) $3. \text{Ca}(\text{OH})_2 + 1. \text{Al}_2(\text{SO}_4)_3 \rightarrow 3. \text{CaSO}_4 + 2. \text{Al}(\text{OH})_3$. Type of reaction: double

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displacement. 3) $3. \text{Mg} + 1. \text{Fe}_2\text{O}_3 \rightarrow 2. \text{Fe} + 3. \text{MgO}$. Type of ...

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Types of Chemical Reaction Worksheet.
Balance the reactions 1 to 6. and
indicate which type of chemical reaction
(synthesis, decomposition, single-

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displacement, double-displacement or combustion) is being represented: $2_ \text{NaBr} + _1_ \text{Ca(OH)}_2 (_1_ \text{CaBr}_2 + _2_ \text{NaOH}$ Reaction Type : Double replacement. $2_ \text{NH}_3 + _1_ \text{H}_2\text{SO}_4 (_1_ (\text{NH}_4)_2\text{SO}_4$ Reaction Type : Synthesis.

Types of Chemical Reaction Worksheet

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Types of Reactions Worksheet - Solutions. Balance the following equations and indicate the type of reaction taking place: 1) $3. \text{NaBr} + 1. \text{H}_3\text{PO}_4 \rightarrow 1. \text{Na}_3\text{PO}_4 + 3. \text{HBr}$ Type of reaction: ... - Answers. Balance these equations! Note to students: Whenever balancing an equation, it is acceptable to leave spaces blank instead of writing

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Balancing Equations Practice Worksheet

Six Types of Chemical Reaction
Worksheet Answers Balance the
following reactions and indicate which of
the six types of chemical reaction are
being represented: a) $2 \text{NaBr} + 1$

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Types of Chemical Reaction Worksheet Practice Answers ...

Balancing requires a lot of practice, knowledge of reactions, formulae, valances, symbols, and techniques. Often, students lose hope and struggle to solve it. If you are struggling as well,

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Six Types of Chemical Reaction
Worksheet Answers Balance the
following reactions and indicate which of
the six types of chemical reaction are
being represented: a) $2 \text{NaBr} + 1 \text{Ca} \dots$

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Balance the reactions a to e and indicate which types of ...

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