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Worksheet 5 Double Replacement Reactions

Worksheet #5: Double-
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In these reactions, all you do is look at the names of the reactants, and "switch partners". Just be sure that the new pairs come out with the positive ion named first, and paired with a negative ion. 1.

aluminum iodide +
mercury(II) chloride \rightarrow

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2. Double

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Double-Replacement
Reactions In these**

...

Worksheet #5: Double-
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that the new pairs
come out with the
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first, and paired with a

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negative ion. 1.

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Double ...

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names of the
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negative ion. 1.
aluminum iodide +
mercury(II) chloride \rightarrow
2.
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' In these reactions, all

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In these reactions, all you do is look at the names of the reactants, and "switch partners". Just be sure that the new pairs come out with the positive ion named first, and paired with a negative ion. 3. 4. 6. 8. 9. 10.

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Double Displacement Reactions Answer Key **Worksheets - Learny Kids**

One example of a double displacement reaction is the reaction of lead (II) nitrate with ... single replacement
4) double replacement
5) acid-base (double replacement) 6)
synthesis 7)
decomposition .

Practice Balancing

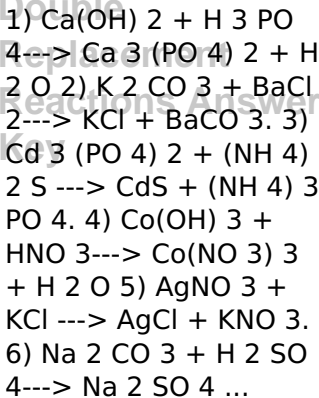
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Chemical Equations
Worksheet . Balance
the following chemical
equations using
coefficients . 1 . 1.
 $\text{Al}(\text{OH})_3 (\text{s}) + 3.$

Unit 5 Chemical Equations and Reactions

Practice Problem
Answers. Back to
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Write correct formulas
for the products in
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ChemTeam: Double Replacement Problem Answers

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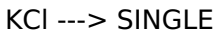
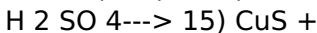
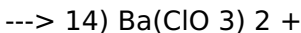
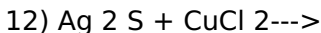
Replacement

Double Displacement

Reactions - Lesson Worksheets

Replacement

Worksheet-Chem 121



REPLACEMENT

REACTION

DESCRIPTION In these

reactions, a free

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element reacts with a compound to form another compound and release one of the elements of the original compound in the elemental state. There are ...

CHEMISTRY REPLACEMENT REACTION WORKSHEET

Single Replacement
Reaction C Synthesis
Reaction A. Double
Replacement Reaction

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D Decomposition
Reaction B. 7. Can two
elements be used as
reactants for a
synthesis reactions? If
yes, give at least one
example from Model 1
to support your
answer. Yes, $4\text{Fe}(s) + 3\text{O}_2(g) \rightarrow 2\text{Fe}_2\text{O}_3(s)$ 8.

Types of Chemical Reactions Key - Google Docs

05 - Worksheet_Double
Displacement Rxn.pdf

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Replacement

05 - Worksheet_Double Displacement

Rxn.pdf

5 Main Patterns of
Reactions * Synthesis *
Decomposition * Single
Displacement * Double
Displacement *
Combustion Synthesis
- To ... Also known as
double replacement;
Usually occurs between
two ionic compounds,
which then in turn

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create two more ionic
compounds.

Types of Reactions - Synthesis, Decomposition, Single and ...

Artwork taken from an unknown source (a worksheet acquired second-hand many years ago). Types of Reactions Chemical reactions can be categorized into a variety of types. Five common types that we

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will study are
synthesis,
decomposition,
combustion, single
replacement and
double replacement.
Be familiar

Types of Reactions - gbschemphys

Classifying Chemical
Reactions Worksheet

Name: _____ Period:

_____ Classify each
reaction as acid/base,
redox, synthesis,
decomposition, single

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Double replacement, double replacement or combustion. They may be more than one. The equations are not balanced. 1.

Classifying Chemical Reactions Worksheet

NaOH. Type of reaction: double replacement. 2) 2.

$\text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow (\text{NH}_4)_2\text{SO}_4$. Type of reaction: synthesis. 3)

4. $\text{C}_5\text{H}_9\text{O} + 27$. Six

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Types of Chemical
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Compilation of the 5
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equations included for
all reactions. UPDATE:
Synthesis Rxn- Word
Equation: Iron(II) +
Sulfur yields Iron(II)
Sulfide Page 8/11.

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